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<b>UTILITY PATENT APPLICATION TRANSMITTAL</b> <small>(Use for new nonprovisional applications under 37 CFR 1.53(b))</small>		<i>Attorney Docket No.</i> 0057-2209-0 <i>First Named Inventor or Application Identifier</i> Kouji HAMANO, et al. <b>Title</b> LITHIUM ION SECONDARY BATTERY AND METHOD OF FABRICATING THEREOF	
<b>APPLICATION ELEMENTS</b> <i>See MPEP chapter 600 concerning utility application contents</i>		<b>Assistant Commissioner for Patents</b> <b>ADDRESS TO:</b> Box Patent Applications Washington, DC 20231	
1. <input checked="" type="checkbox"/> Fee Transmittal Form <i>(Submit an original, and a duplicate for fee processing)</i>  2. <input checked="" type="checkbox"/> Specification [Total Pages 35]  3. <input checked="" type="checkbox"/> Drawing(s) [Total Sheets 7]  4. Oath or Declaration [Total Pages 5] <ul style="list-style-type: none"> <li>a. <input checked="" type="checkbox"/> Newly executed (original)</li> <li>b. <input type="checkbox"/> Copy from a prior application (37 CFR 1.63(d))  <i>(for continuation/divisional with Box 17 completed)</i>  <small>[Note Box 5 below]</small> <ul style="list-style-type: none"> <li>i. <input type="checkbox"/> <u>DELETION OF INVENTOR(S)</u>            Signed statement attached deleting inventor(s) named in the prior application,            see 37 CFR 1.63(d)(2) and 1.33(b).</li> </ul> </li> </ul> 5. <input type="checkbox"/> Incorporation By Reference <i>(useable if Box 4b is checked)</i> The entire disclosure of the prior application, from which a copy of the oath or declaration is supplied under Box 4b, is considered as being part of the disclosure of the accompanying application and is hereby incorporated by reference therein.		<b>ACCOMPANYING APPLICATION PARTS</b> <ul style="list-style-type: none"> <li>6. <input type="checkbox"/> Assignment Papers (cover sheet &amp; document(s))</li> <li>7. <input type="checkbox"/> 37 CFR 3.73(b) Statement <input type="checkbox"/> Power of Attorney <i>(when there is an assignee)</i></li> <li>8. <input type="checkbox"/> English Translation Document <i>(if applicable)</i></li> <li>9. <input type="checkbox"/> Information Disclosure <input type="checkbox"/> Copies of IDS Citations Statement (IDS)/PTO-1449</li> <li>10. <input type="checkbox"/> Preliminary Amendment</li> <li>11. <input checked="" type="checkbox"/> White Advance Serial No. Postcard</li> <li>12. <input type="checkbox"/> Small Entity <input type="checkbox"/> Statement filed in prior app, Statement(s) Status still proper and desired</li> <li>13. <input type="checkbox"/> Certified Copy of Priority Document(s)</li> <li>14. <input checked="" type="checkbox"/> Other: Notice of Priority</li> </ul>	
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16. Amend the specification by inserting before the first line the sentence: This application is a <input type="checkbox"/> Continuation <input type="checkbox"/> Division <input type="checkbox"/> Continuation-in-part (CIP) of application Serial No. ?, filed on ?.			
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IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

Inventors: Kouji HAMANO, et al.

Serial No: NEW APPLICATION

Filing Date: HEREWITH

Title: LITHIUM ION SECONDARY BATTERY AND METHOD OF FABRICATING  
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Washington, D.C. 20231

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Respectfully submitted,

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Docket No. 0057-2209-0

IN THE UNITED STATES PATENT AND TRADEMARK OFFICE

IN RE APPLICATION OF: Kouji HAMANO, et al.

GROUP ART UNIT:

SERIAL NUMBER: NEW APPLICATION

EXAMINER:

FILED: HEREWITH

FOR: LITHIUM ION SECONDARY BATTERY AND METHOD OF FABRICATING THEREOF

REQUEST FOR PRIORITY UNDER 35 U.S.C. 119  
AND THE INTERNATIONAL CONVENTION

Assistant Commissioner for Patents  
Washington, D.C. 20231

Sir:

In the matter of the above-identified application for patent, notice is hereby given that the applicants claim as priority:

<u>COUNTRY</u>	<u>APPLICATION NO:</u>	<u>MONTH/DAY/YEAR</u>
Japan	8-324032	December 4, 1996

A Certified copy of the corresponding Convention Application will be submitted prior to payment of the Final Fee.

Respectfully submitted,

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**Lithium Ion Secondary Battery  
and Method of fabricating Thereof**

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**Background of the Invention**

**1. Field of the invention**

This invention relates to a lithium ion secondary battery in which positive and negative electrodes face each other sandwiching a separator which keeps an electrolytic solution, more particularly to a battery structure in which electric connection between a positive electrode and a negative electrode (electrodes) and separator is improved so that it can be made into thin form and the like optional forms and to a production method for the formation of said structure.

15    **2. Discription of the Related Art**

There is a growing demand for the miniaturization and lightening of portable electronic instruments, and it is essential to improve performance of batteries to meet such a demand. Because of this, development and improvement of various batteries have been attempted in recent years with the aim of improving the battery performance. Expected characteristics of batteries to be improved include high voltage, large energy density, tolerance for large load resistance, optional shaping, safety and the like. Particularly, lithium ion battery is a secondary battery which can realize the highest voltage, largest

energy density and tolerance for largest load resistance among existing batteries, and its improvement is still being made actively.

As its main composing elements, the lithium ion secondary battery has a positive electrode, a negative electrode and an ion conducting layer inserted between these electrodes. In the lithium ion secondary batteries which have been put into practical use, a plate-shaped material prepared by mixing powder of a lithium-cobalt oxide or the like active material with powder of an electron conducting substance and a binder resin and coating the mixture on an aluminum collector is used as the positive electrode, and another plate-shaped material prepared by mixing powder of a carbonaceous active material with a binder resin and coating the mixture on a copper collector is used as the negative electrode. Also, a porous film such as of polyethylene, polypropylene or the like filled with a lithium ion-containing non-aqueous solution is used as the ion conducting layer.

For example, Fig. 7 is a sectional view showing the structure of a prior art cylindrical lithium ion secondary battery disclosed in JP-A-8-83608 (the term "JP-A" as used herein means an "unexamined published Japanese patent application"). In Fig. 7, 1 is an armor case made of stainless steel or the like which also serves as a negative electrode terminal, 2 is an electrode body contained in the armor case 1, and the electrode body 2 has a structure in which a positive electrode 3, a separator 4 and

a negative electrode 5 are coiled in a spiral shape. In order to maintain electric connection among the positive electrode 3, separator 4 and negative electrode 5, it is necessary to apply external pressure to surfaces of the electrode body 2. Because 5 of this, contact among all surfaces is maintained by putting the electrode body 2 inserted into a strong metal case. In the case of a square battery, strips of electrode body are tied up into a bundle and put into a square metal case, thereby pressing them with external force.

10 As described in the foregoing, in the currently available lithium ion secondary batteries, strong armor cases made of metals and the like are used as a means to closely adhere positive and negative electrodes. Without the armor case, the electrodes are peeled off, so that it becomes difficult to maintain electric 15 connection between the electrodes via an ion conducting layer (separator) and the battery characteristics therefore are deteriorated. On the other hand, not only the energy density of the battery itself is reduced because of the large weight and volume of the armor case occupying entire portion of the battery, 20 but also shapes of the battery are limited due to rigidity of the armor case itself, thus causing a difficulty in making optional shapes.

In view of such backgrounds, development of a lithium ion secondary battery which does not require a strong armor case has 25 been attempted with the aim of achieving lightening and thinning

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the battery. The point of the development of such a armor case-free battery is how to maintain electric connection of a positive electrode, a negative electrode and an ion conducting layer (separator) which is sandwiched by them, without applying 5 external force. As such a connecting means which does not require external force, a method has been proposed in which the electrodes are closely adhered to the separator making use of a resin or the like.

For example, JP-A-5-159802 discloses a production method 10 in which an ion conductive solid electrolyte layer and positive and negative electrodes are integrated into one body by their heat treatment using a thermoplastic resin binder. In this case, the electrodes are closely adhered to each other by integrating the electrodes and the electrolyte layer into one body, so that 15 electric connection between the electrodes is maintained and the integrated body functions as a battery without applying external force.

Since the prior art lithium ion secondary batteries are constructed in the aforementioned manner, a battery which uses 20 a strong armor case to ensure adhesiveness between electrodes and a separator and electric connection between electrodes is disadvantageous in producing a battery having large energy density, because the ratio of volume and weight of the non-electricity generating part armor case to the entire battery 25 portion becomes large. Also, though a method in which electrodes

are closely adhered to an ion conducting body via an adhesive resin has been proposed, it causes a problem in that ionic conduction resistance inside the battery cell increases and the battery characteristics are reduced due to large resistance of the  
5 adhesive resin layer, when a solid electrolyte layer is closely adhered to the electrodes simply via the adhesive resin.

In addition, in the case of the battery of JP-A-5-159802, the electrodes are bonded to a solid electrolytic layer with a binder, but since interfaces of the electrodes and the  
10 electrolytic layer are covered with the binder, it is disadvantageous in terms of ionic conductivity when compared for example with a case in which liquid electrolytes are used. Even if a binder having ionic conductivity is used, a material having an ionic conductivity equal to or larger than that of liquid  
15 electrolytes is not generally known, so that it causes a problem in that battery performance similar to that of a battery in which liquid electrolytes are used cannot be obtained easily.

20

#### Summary of the Invention

The present invention has been accomplished as a result of intensive studies on the suitable bonding method of a separator and electrodes, conducted by the present inventors with the aim of resolving the aforementioned problems, and it contemplates  
25 providing a lithium ion secondary battery having excellent charge

and discharge characteristics, which can closely adhere the electrodes to the separator without increasing ionic conduction resistance between the electrodes and without using a strong armor case, so that it has large energy density and can be made into  
5 a thin and optional shape.

A first aspect of the lithium ion secondary battery of the present invention is a battery which comprises a positive electrode having a positive electrode collector and a positive electrode active material layer formed on the positive electrode collector ; a negative electrode having a negative electrode collector and a negative electrode active material layer formed on the negative electrode collector, a separator which is arranged between the positive electrode and negative electrode and keeps  
10 a lithium ion-containing electrolytic solution and a porous adhesive resin layer which bonds the positive electrode active material layer and the negative electrode active material layer to the separator and keeps the electrolytic solution to mutually connect the positive electrode, separator and negative electrode  
15 electrically.  
20

A second aspect of the lithium ion secondary battery of the present invention is the battery according to the first aspect wherein hole ratio of the porous adhesive resin layer is equal to or larger than that of the separator.

25 A third aspect of the lithium ion secondary battery of

the present invention is the battery according to the first aspect wherein hole ratio of the porous adhesive resin layer is more than 35%.

A fourth aspect of the lithium ion secondary battery of  
5 the present invention is the battery according to the first aspect wherein the ionic conduction resistivity of the adhesive resin layer in which the electrolytic solution is kept is set to a value equal to or lower than the ionic conduction resistivity of the separator in which the electrolytic solution is kept.

10 A fifth aspect of the lithium ion secondary battery of the present invention is the battery according to the first aspect wherein the bonding strength between the positive electrode active material layer and the separator is set to a value equal to or larger than the bonding strength between the positive  
15 electrode active material layer and the positive electrode collector, and the bonding strength between the negative electrode active material layer and the separator is set to a value equal to or larger than the bonding strength between the negative electrode active material layer and the negative electrode collector.  
20

A sixth aspect of the lithium ion secondary battery of the present invention is the battery according to the first aspect wherein the adhesive resin layer is made of a fluoride resin or a mixture which uses a fluoride resin as the main component.

25 A seventh aspect of the lithium ion secondary battery of

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the present invention is the battery according to the sixth aspect wherein polyvinylidene fluoride is used as the fluoride resin.

An eighth aspect of the method of the present invention for fabricating a lithium ion secondary battery, is the method which comprises the steps of: preparing a positive electrode by forming a positive electrode active material layer on a positive electrode collector; preparing a negative electrode by forming a negative electrode active material layer on a negative electrode collector; preparing an adhesive resin solution, by dispersing a fluoride resin or a mixture containing a fluoride resin as the main component in N-methylpyrrolidone; coating the adhesive resin solution to at least one of the surface of the positive electrode active material layer and the facing surface of the separator and to at least one of the surface of the negative electrode active material layer and the facing surface of the separator; fitting the positive electrode active material layer and the negative electrode active material layer upon respective surfaces of the separator; evaporating the N-methylpyrrolidone from the adhesive resin solution to form porous adhesive resin layers so as to bond the positive electrode active material layer and the negative electrode active material layer upon respective surfaces of the separator to form a laminated body; and supplying a lithium ion-containing electrolytic solution to the laminated body.

A ninth aspect of the method of fabricating a lithium ion secondary battery according to the eighth aspect is the method

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wherein the adhesive resin solution contains 3-10 parts by weight of a fluoride resin or a mixture containing a fluoride resin as the main component in N-methylpyrrolidone.

5 A tenth aspect of the method of fabricating a lithium ion secondary battery according to the eighth aspect wherein the step of evaporating comprises a step of heating.

A eleventh aspect of the method of fabricating a lithium ion secondary battery according to the eighth aspect is the method wherein the step of coating comprises a step of coating by using  
10 a bar coator.

A twelfth aspect of the method of fabricating a lithium ion secondary battery according to the eighth aspect is the method wherein the step of coating comprises a step of coating by using a spray gun.

15 A thirteenth aspect of the method of fabricating a lithium ion secondary battery according to the eighth aspect is the method wherein the step of coating comprises a step of dipping the separator in an emulsified solution of the adhesive resin and then pulling it up.

20

#### Brief Description of the Drawings

Fig. 1 is a sectional view showing the battery structure, or electrode laminate, of the lithium ion secondary battery of  
25 the first aspect of the present invention.

Fig. 2 is an explanation drawing showing a method for coating an adhesive resin solution by a bar coat method as a practical mode of the present invention.

5 Fig. 3 is an explanation drawing showing a method for coating an adhesive resin solution using a spray gun as a practical mode of the present invention.

Fig. 4 is an explanation drawing showing a method for coating an adhesive resin solution by a dipping process as a practical mode of the present invention.

10 Fig. 5 is a sectional illustration showing the lithium ion secondary battery according to the first aspect of the present invention.

15 Fig. 6 is a characteristic drawing showing a relationship between the internal resistance and the amount of adhesive resin in the adhesive resin solution at the time of the formation of the adhesive resin layer according to the first aspect of the present invention.

Fig. 7 is a sectional illustration showing an example of the prior art lithium ion secondary battery.

20 In these drawings, 3 is a positive electrode, 4 is a separator, 5 is a negative electrode, 6 is a positive electrode collector, 7 is a positive electrode active material layer, 9 is a negative electrode active material layer, 10 is a negative electrode collector, 11 is an adhesive resin layer and 12 is  
25 through holes.

Description of the preferred Embodiment

Fig. 1 is a sectional view showing battery structure,  
5 namely structure of the electrode laminate, of the lithium ion  
secondary battery of the first aspect of the present invention,  
in which 3 is a positive electrode prepared by bonding a positive  
electrode active material layer 7 to a positive electrode  
collector 6, 5 is a negative electrode prepared by bonding a  
10 negative electrode active material layer 9 to a negative electrode  
collector 10, 4 is a separator which is arranged between the  
positive electrode 3 and negative electrode 5 and keeps a lithium  
ion-containing electrolytic solution and 11 is a porous adhesive  
resin layer which bonds the positive electrode active material  
15 layer 7 and negative electrode active material layer 9 to the  
separator 4 and has a large number of through holes 12 that  
communicate the positive electrode active material layer 7 and  
negative electrode active material layer 9 with the separator 4.

Since the electrode layers (namely active material layers  
20 7 and 9) and the separator 4 which becomes an electrolyte layer  
are mutually bonded by the porous adhesive resin layer 11, bonding  
strength between the electrodes and the separator is ensured and  
it becomes possible to prevent peeling between the electrodes and  
separator, which was difficult to achieve in the conventional  
25 batteries. Also, an electrolytic solution is kept in the through

holes 12 which are formed in the inside part, namely in the adhesive resin layer 11, and communicate interfaces of the electrodes and separator, so that excellent ionic conductivity of the electrode-electrolyte interface can be ensured and ionic conduction resistance between the electrodes can be reduced.

Coming and going quantity of ions in the active material layers inside the electrodes and drift speed and drift quantity of ions into facing electrodes can be controlled to the same levels of the prior art lithium ion batteries which use armor cases.

Electric connection between the electrodes can be maintained without applying external force. In consequence, a strong armor case is not required for keeping the battery structure, so that lightening and thinning of the battery can be made, optional shapes can be formed and excellent charge and discharge characteristics and battery performance can be obtained which are similar to those of the conventional batteries that use an electrolytic solution.

Also, when the ionic conduction resistivity of the adhesive resin layer 11 in which the electrolytic solution is kept is set to a value equal to or lower than the ionic conduction resistivity of the separator 4 in which the electrolytic solution is kept, the adhesive resin layer 11 prevents deterioration of charge and discharge characteristics. The charge and discharge characteristics as a battery can be maintained at the level of prior art batteries.

The ionic conduction resistivity of the adhesive resin layer 11 can be adjusted mainly by changing its hole ratio and thickness. The hole ratio can be controlled for example by changing the ratio of an adhesive resin to N-methylpyrrolidone 5 in an adhesive resin solution which forms the adhesive resin layer.

It is desirable also to set the bonding strength between the positive electrode active material layer and the separator to a value equal to or larger than the bonding strength between 10 the positive electrode active material layer and the positive electrode collector, and the bonding strength between the negative electrode active material layer and the separator to a value equal to or larger than the bonding strength between the negative electrode active material layer and the negative 15 electrode collector, namely to set the bonding strength to a value equal to or larger than the bonding strength of the adhered and integrated body of the active material layers and collectors inside the battery. A peeling test carried out after formation 20 of a battery confirmed that, when bonding strength between the electrodes and the separator is sufficiently large, destruction of the electrodes (peeling of the active material layers from the collectors) occurs taking preference over peeling between the electrodes and the separator. This bonding strength can be controlled for example by changing thickness of the adhesive resin 25 layer and selecting proper adhesive resin.

25

With regard to the adhesive resin to be used in the adhesive resin layer 11 for the adhesion of the active material layers to the separator, a resin which is insoluble in the electrolytic solution, does not generate electrochemical reactions in the battery and becomes porous membrane, namely a fluoride resin or a mixture which contains a fluoride resin as the main component, is used. Its illustrative examples include vinylidene fluoride, 4-ethylene fluoride and the like polymers which has fluorine atom in its molecular structure, or their mixtures with methyl polymethacrylate, polystyrene, polyethylene, polypropylene and the like. Of these resins, polyvinylidene fluoride as a fluoride resin is particularly desirable.

15

The lithium ion secondary battery constructed in the aforementioned manner is produced by adhering an adhesive resin solution, prepared by dispersing a fluoride resin or a mixture containing a fluoride resin as the main component in N-methylpyrrolidone, to both sides of the separator 4, thereby laminating the positive electrode active material layer 7 and negative electrode active material layer 9, and subsequently evaporating N-methylpyrrolidone to effect formation of the porous adhesive resin layer 11 which bonds the positive electrode active material layer 7 and the negative electrode active material layer 9 to the separator 4.

25

With regard to the active materials to be used in the

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present invention, examples of the material to be used in the positive electrode include complex oxides of lithium with cobalt, nickel, manganese and the like transition metals, chalcogen compounds or complex compounds thereof and these compounds which further contain various additional elements, and examples of the material to be used in the negative electrode include easy-graphitized carbon, hard-graphitized carbon, polyacene, polyacetylene and the like carbonaceous compounds and pyrene, perylene and the like acene structure-containing aromatic hydrocarbon compounds, but any other material may also be used with the proviso that it can occlude and release lithium ions which become the core of the battery actuation. In addition, these active materials are used in a granular form, and their useful particle size is 0.3 to 20  $\mu\text{m}$ , most preferably 0.3 to 5  $\mu\text{m}$ .

As the binder resin which is used for making the active materials into electrode plates, any resin can be used, provided that it is insoluble in the electrolytic solution and does not generate electrochemical reactions in the electrode laminate. Its illustrative examples include vinylidene fluoride, ethylene fluoride, acrylonitrile, ethylene oxide and the like homopolymers or copolymers and ethylene propylene diamine rubber.

Any metal which is stable in the battery can be used as the collector, but aluminum is preferably used in the positive electrode and copper in the negative electrode. With regard to the shape of the collector, foil, net, expanded metal and the like

shapes can be employed, but net, expanded metal or the like shape having a large void area is desirable in terms of its feasibility to keep the electrolytic solution after bonding.

With regard to the adhesive resin to be for the adhesion of the collectors and electrodes, a resin which is insoluble in the electrolytic solution, does not generate electrochemical reactions in the battery and becomes porous membrane is used similar to the case of the adhesive resin which is used for the adhesion of the electrodes and separator. Its illustrative examples include vinylidene fluoride, 4-ethylene fluoride and the like polymers which has fluorine atom in its molecular structure, or their mixtures with methyl polymethacrylate, polystyrene, polyethylene, polypropylene and the like.

Also, any material can be used as the separator, with the proviso that it is electron insulating material having sufficient strength, such as a porous membrane, a net, non-woven fabric or the like. Though its construction material is not particularly limited, polyethylene or polypropylene is desirable from the viewpoint of adhesiveness and safety.

With regard to the solvent and electrolyte salt which are included in the electrolytic solution to be used as the ionic conduction body, non-aqueous solvents and lithium-containing electrolyte salts used in conventional batteries can be used.

Illustrative examples of the solvent include dimethoxyethane, diethoxyethane, diethyl ether, dimethyl ether and the like ether

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solvents and propylene carbonate, ethylene carbonate, diethyl carbonate, dimethyl carbonate and the like ester solvents, which may be used as a single solution or as a mixture solution of two of the same or different solvents. Illustrative examples of the  
5 electrolyte salt to be used in the electrolytic solution include LiPF<sub>6</sub>, LiAsF<sub>6</sub>, LiClO<sub>4</sub>, LiBF<sub>4</sub>, LiCF<sub>3</sub>SO<sub>3</sub>, LiN(CF<sub>3</sub>SO<sub>2</sub>)<sub>2</sub>, LiC(CF<sub>3</sub>SO<sub>2</sub>)<sub>3</sub>, LiN(C<sub>2</sub>F<sub>5</sub>SO<sub>2</sub>)<sub>2</sub> and the like.

In addition, a method in which a bar coater is used, a method in which a spray gun is used or a dipping process is used  
10 as a means for coating the adhesive resin.

For example, as shown in the explanation illustration of Fig. 2, the method in which a bar coater is used is carried out by adding dropwise an adhesive resin in lines to a separator material 13 which is moving, rolling the added resin with a bar  
15 coater 14, thereby effecting uniform coating of the adhesive resin on the entire area of one side of the separator material, and then, after coating of the side, the separator material is twisted at an angle of 180 degree (reversed) and the adhesive resin is coated on the un-coated side in the same manner. In this connection,  
20 14 is an outlet for use in the dropwise addition of the adhesive resin, 16 is a backup roll and 17 is a separator roll.

As shown in the explanation illustration of Fig. 4, in the method in which a spray gun is used, an adhesive resin solution or a liquid adhesive resin is packed in a spray gun 18 and then  
25 the adhesive resin solution or liquid adhesive resin is sprayed

on the separator material 13, thereby effecting adhesion of the adhesive resin on the separator material 13. By arranging at least one spray gun 18 on each side of the separator material 13 and continuously spraying the adhesive resin solution while 5 moving the separator material 13, the adhesive resin can be coated continuously on both sides of the separator. Similar to the case of the bar coater method, the adhesive resin can be coated on the separator within a short period of time and in a large amount.

Also, the dipping process is a coating method in which 10 a separator is dipped in an emulsified solution of an adhesive resin and then pulled up, thereby effecting coating of the adhesive resin on both sides of the separator. That is, as shown in Fig. 4, entire part of the separator material 13 is dipped in an emulsified solution of adhesive resin (to be referred to as 15 adhesive resin emulsion hereinafter) 21 and then pulled up while removing excess adhesive resin emulsion 21 by holding it between excess liquid removing rolls 24, thereby effecting coating of the adhesive resin on the whole sides of the separator. In the drawing, 20 22 is a backup roll. By the use of the dipping process, the coating step can be simplified and the adhesive resin can be coated within a short period of time and in a large amount.

#### Embodiments

Embodiments of the present invention are given below by way of illustration and not by way of limitation.

25 **Embodiment 1**

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Firstly, preparation of a positive electrode is described.

A positive electrode active material paste prepared by dispersing 87 parts by weight of LiCoO<sub>2</sub>, 8 parts by weight of graphite powder and 5 parts by weight of polyvinylidene fluoride in N-methylpyrrolidone was coated in a thickness of 300 µm by a doctor blade method to form an active material thin film. An aluminum net having a thickness of 30 µm to be used as a positive electrode collector was put on the film, on which was again coated the positive electrode active material paste prepared to a thickness of 300 µm by the doctor blade method. This was made into a half-dried state by allowing it to stand for 60 minutes in an oven of 60°C. The thus prepared laminate was closely adhered by lightly rolling it using a spin roller whose roll clearance has been adjusted to 550 µm, thereby preparing a positive electrode.

The thus prepared positive electrode was soaked in an electrolytic solution and then peel strength of the positive electrode active material layer and the positive electrode collector was measured to find a value of 20 to 25 gf/cm.

Next, preparation of a negative electrode is described.

A negative electrode active material paste prepared by dispersing 95 parts by weight of Meso Phase Microbeads Carbon (trade name: manufactured by Osaka Gas) and 5 parts by weight of polyvinylidene fluoride in N-methylpyrrolidone (to be referred

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to as NMP hereinafter) was coated in a thickness of 300  $\mu\text{m}$  by the doctor blade method to form an active material thin film. A copper net having a thickness of 20  $\mu\text{m}$  to be used as a negative electrode collector was put on the film, on which was again coated the 5 negative electrode active material paste prepared to a thickness of 300  $\mu\text{m}$  by the doctor blade method. This was made into a half-dried state by allowing it to stand for 60 minutes in an oven of 60°C. The thus prepared laminate was closely adhered by lightly rolling it using a spin roller whose roll clearance has 10 been adjusted to 550  $\mu\text{m}$ , thereby preparing a negative electrode.

The thus prepared negative electrode was soaked in an electrolytic solution and then peel strength of the negative electrode active material layer and the negative electrode collector was measured to find a value of 10 to 15 gf/cm.

15           The following describes preparation of an electrode laminate.

Firstly, a viscous adhesive resin solution was prepared by mixing 5 parts by weight of polyvinylidene fluoride with 95 parts by weight of N-methylpyrrolidone (to be referred to as NMP 20 hereinafter) at that compositional ratio and thoroughly stirring the mixture until it became uniform solution.

Next, the thus prepared adhesive resin solution was coated on both sides of a separator material to be used as a separator, namely a porous polypropylene sheet (trade name Cell 25 Guard #2400, manufactured by Hoechst).

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Coating of the adhesive resin was carried out by the bar coater method.

A rolled-up separator material 13, namely a porous polypropylene sheet (trade name Cell Guard #2400, manufactured 5 by Hoechst) having a width of 12 cm and a thickness of 25 µm, was took out and the adhesive resin solution was added dropwise to one side of the sheet in a line of orthogonal direction to the take out direction of the separator material. It was able to coat the adhesive resin solution thus applied in a line uniformly on 10 whole sides of the separator material 13, by spinning a bar coater 15 prepared by minutely coiling a filament of 0.5 mm in diameter around a tube of 1 cm in diameter, while simultaneously moving the separator material 13. In this connection, coating mass of the adhesive resin can be controlled by changing amount of the 15 adhesive resin solution to be added dropwise.

Thereafter, and before the adhesive resin solution was dried, the positive electrode and the negative electrode were closely adhered and laminated in such a way that they faced each other sandwiching the separator, and the thus pasted electrode 20 laminate was allowed to stand for 2 hours in a hot air oven of 60°C to effect evaporation of NMP, thereby preparing an electrode laminate. By the evaporation of NMP from the adhesive resin layer, it becomes a porous adhesive resin layer which has through holes that communicate the positive electrode and negative electrode 25 with the separator.

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Subsequently, an electrolytic solution prepared using ethylene carbonate and diethyl carbonate as the solvent and LiPF<sub>6</sub> as the solute was injected into the electrode laminate. When peel strengths of the positive electrode active material layer and the 5 separator and of the negative electrode active material layer and the separator were measured at this stage, the strengths were found to be 25 to 30 gf/cm and 15 to 20gf/cm, respectively. A lithium ion battery was completed by packing the thus electrolytic solution-injected electrode laminate with a resin laminated aluminum film(to be referred to as an aluminum laminate film 10 hereinafter) and sealing the opening by hot melting.

Fig. 5 is a sectional illustration showing the lithium ion secondary battery of the first practical mode 1 of this invention. In this drawing, 26 is an armor aluminum laminate pack 15 and 8 is an electrode laminate sealed in the armor aluminum laminate pack. The electrode laminate 8 is composed of a positive electrode 3, a separator 4 and a negative electrode 5. An adhesive resin layer 11 is arranged between the positive electrode 3 and separator 4 and the negative electrode 5 and separator 4, thereby bonding the positive electrode 3 and negative electrode 5 to the separator 4 tightly. An electrolytic solution is kept in through holes 12 in the adhesive resin layer 11, holes in the electrode active material layers and holes in the separator. 20

As described above, in this lithium ion secondary battery, 25 the positive electrode 3 and separator 4 and the negative

electrode 5 and separator 4 are closely adhered by the adhesive resin layer 11, through holes 12 which communicate between the electrodes 3 and 5 and the separator 4 are formed in a large number in the adhesive resin layer 11 and an electrolytic solution is  
5 kept in the through holes 12 to ensure excellent ionic conductivity, so that it was able to produce a thin and light battery which has excellent charge and discharge characteristics and does not require external pressurization, namely which does not require a strong armor case.

10 The characteristic graph of Fig. 6 shows internal resistance of a battery when an adhesive resin layer was formed by changing the amount of polyvinylidene fluoride resin as an adhesive resin in an adhesive resin solution to 5 parts by weight, 7 parts by weight or 10 parts by weight based on NMP. It can be  
15 seen that the resistance increases sharply between 5 parts by weight and 7 parts by weight. Since thickness of the adhesive resin layer 11 was proportional to the amount of adhesive layer in the adhesive resin solution, it was considered that the resistance increased rapidly in this range due to sudden changes  
20 in the retention of the electrolytic solution and distribution conditions of the electrolytic solution in the adhesive resin layer 11. In this connection, the resistance value in the case of 5 parts by weight was almost the same as the resistance value which was measured by not arranging the adhesive resin layer 11  
25 but applying sufficient pressure to the surfaces between the

electrodes 3 and 5 and the separator 4.

An adhesive resin solution preferably contains 3-10 parts by weight of a fluoride resin or a mixture containing a fluoride resin as the main component in N-methylpyrrolidone in the view 5 of the ion conductivity and bonding strength of the adhesive layer. More preferably 4-6 parts by weight. And a thickness of the adhesive resin layer is preferably less than  $0.5\mu\text{m}$ . Not only polyvinylidene fluoride resin but also other adhesive resin can be used as an adhesive resin as same, and come to the 10 same result.

#### Embodiment 2

In preparing the adhesive resin layer 11 shown in Embodiment 1, viscous adhesive resin solutions were prepared by 15 mixing N-methylpyrrolidone with each of the following compounds 1-9 in stead of polyvinylidene fluoride at the same compositional ratio as Embodiment 1 respectively.

1. Polytetrafluoroethylene
2. Copolymer of vinylidene fluoride and acrylonitrile
- 20 3. Mixture of polyvinylidene fluoride and polyacrylonitrile
4. Mixture of polyvinylidene fluoride and polyethylene oxide
5. Mixture of polyvinylidene fluoride and polyethylene terephthalate
6. Mixture of polyvinylidene fluoride and methyl polymethacrylate
- 25 7. Mixture of polyvinylidene fluoride and polystyrene

8.Mixture of polyvinylidene fluoride and polypropylene

9.Mixture of polyvinylidene fluoride and polyethylene

Using these adhesive resin solutions, the electrode laminate 8 was prepared by the same method of Embodiment 1. When 5 peel strengths of the positive electrode active material layer and separator and of the negative electrode active material layer and separator in this electrode laminate 8 were measured, the strengths were converged at 25 to 70 gf/cm and 15 to 70 gf/cm, respectively. Thereafter, keeping of an electrolytic solution, 10 packing of the battery with an aluminum laminate film and sealing of the opening were carried out in the same manner as described in Embodiment 1, thereby obtaining respective lithium ion secondary batteries. Similar to the case of Inventive Embodiment 1, thin and light batteries having excellent charge and discharge 15 characteristics were obtained.

Though a case in which an adhesive resin solution is coated by a bar coater method was shown in the above embodiments, the adhesive resin solution may also be coated using a spray gun.

As shown in Fig. 3, a rolled-up separator material, namely 20 a porous polypropylene sheet (trade name Cell Guard #2400, manufactured by Hoechst) having a width of 12 cm and a thickness of 25  $\mu\text{m}$ , was took out and an adhesive resin solution was sprayed to the separator using a spray gun which has been packed with the adhesive resin solution. By the spraying, it was able to coat 25 the adhesive resin solution uniformly on both sides of the

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separator material. Also, it was able to adjust coating mass of the adhesive resin solution by changing the splaying quantity.

Adhesion of the adhesive resin solution may also be effected by a dipping process.

5 A rolled-up separator material, namely a porous polypropylene sheet (trade name Cell Guard #2400, manufactured by Hoechst) having a width of 12 cm and a thickness of 25 µm, was took out and then dipped in a solution which has been prepared by mixing the adhesive resin solution shown in Embodiments 1 and  
10 2 with hexane at a weight ratio of 1:1. This solution is in an emulsified state. Thereafter, when the separator material was put between excess liquid removing rolls to remove excess liquid and then air was blew against the separator to evaporate hexane, it was confirmed that the adhesive resin layer was uniformly  
15 adhered to both sides of the separator.

In the first aspect of the lithium ion secondary battery of the present invention, the battery is equipped a positive electrode prepared by bonding a positive electrode active material layer to a positive electrode collector, a negative electrode prepared by bonding a negative electrode active material layer to a negative electrode collector, a separator which is arranged between the positive electrode and negative electrode and keeps a lithium ion-containing electrolytic solution and a porous adhesive resin layer which bonds the  
25

PCT/JP2008/050006

positive electrode active material layer and the negative electrode active material layer to the separator and keeps the electrolytic solution to mutually connect the positive electrode, separator and negative electrode electrically, so that the  
5 electrodes can be adhered closely to the separator by the adhesive resin layer and proper ionic conductivity of the electrode-electrolyte interface can be secured by the electrolytic solution kept in the adhesive resin layer through holes which communicate between the electrodes and the separator, thus exerting an effect  
10 of being able to obtain a lithium ion secondary battery which has excellent charge and discharge characteristics, can be made into a thin form having large energy density and is able to be formed into optional shapes.

In the second aspect of the lithium ion secondary battery  
15 of the present invention, hole ratio of the porous adhesive resin layer is equal to or larger than that of the separator, so that both ionic conductivity of the electrode-electrolyte interface and bonding strength are kept more appropriately.

In the third aspect of the lithium ion secondary battery  
20 of the present invention, hole ratio of the porous adhesive resin layer is more than 35%, so that both ionic conductivity of the electrode-electrolyte interface and bonding strength are kept more appropriately.

In the fourth aspect of the lithium ion secondary battery  
25 of the present invention, the ionic conduction resistivity of the

adhesive resin layer in which electrolytic solution is kept according to the first aspect is set to a value equal to or lower than the ionic conduction resistivity of the electrolytic solution-keeping separator, so that excellent charge and discharge characteristics can be maintained without deteriorating the charge and discharge characteristics.

In the fifth aspect of the lithium ion secondary battery of the present invention, in accordance with the first aspect, the bonding strength between the positive electrode active material layer and the separator is set to a value equal to or larger than the bonding strength between the positive electrode active material layer and the positive electrode collector, and the bonding strength between the negative electrode active material layer and the separator is set to a value equal to or larger than the bonding strength between the negative electrode active material layer and the negative electrode collector, so that destruction of the electrodes occurs taking preference over peeling between the electrodes and the separator. The electrodes can be adhered to the separator sufficiently strongly by the adhesive resin layer. Electric connection between the electrodes can be fully maintained.

In the sixth aspect of the lithium ion secondary battery of the present invention, a fluoride resin or a mixture which uses a fluoride resin as the main component is used as the adhesive resin layer in the first aspect, so that a lithium ion secondary

battery having the aforementioned excellent characteristics can be obtained.

In the seventh aspect of the lithium ion secondary battery of the present invention, polyvinylidene fluoride is used as the 5 fluoride resin of the fifth aspect, so that a lithium ion secondary battery having more excellent characteristics can be obtained.

In the method of the present invention for the production of a lithium ion secondary battery, an adhesive resin solution prepared by dispersing a fluoride resin or a mixture containing 10 a fluoride resin as the main component in N-methylpyrrolidone is adhered to both sides of a separator which is arranged between a positive electrode prepared by bonding a positive electrode active material layer to a positive electrode collector and a negative electrode prepared by bonding a negative electrode 15 active material layer to a negative electrode collector, thereby laminating the positive electrode active material layer and negative electrode active material layer, and then the N-methylpyrrolidone is evaporated to effect formation of a porous adhesive resin layer which bonds the positive electrode active 20 material layer and the negative electrode active material layer to the separator, thus exerting an effect of being able to obtain a lithium ion secondary battery easily with good workability, which has excellent charge and discharge characteristics, can be made into a thin form having large energy density and is able to 25 be formed into optional shapes.

And by using a adhesive resin solution contains 3-10 parts by weight of a fluoride resin or a mixture containing a fluoride resin as the main component in N-methylpyrrolidone, both of ion conductivity and bonding strength of the adhesive resin layer are  
5 kept more appropriately.

**What is claimed is:**

1           1. A lithium ion secondary battery comprising:  
2                 a positive electrode having a positive electrode  
3             collector and a positive electrode active material layer formed  
4             on said positive electrode collector ;  
5                 a negative electrode having a negative electrode  
6             collector and a negative electrode active material layer formed  
7             on said negative electrode collector;  
8                 a separator which is arranged between said positive  
9             electrode and negative electrode and keeps a lithium ion-  
10           containing electrolytic solution; and  
11                 a porous adhesive resin layer which bonds said  
12           positive electrode active material layer and said negative  
13           electrode active material layer to said separator and keeps said  
14           electrolytic solution to mutually connect said positive electrode,  
15           separator and negative electrode electrically.

1           2. The lithium ion secondary battery according to claim  
2           1 wherein hole ratio of said porous adhesive resin layer is equal  
3           to or larger than that of the separator.

1           3. The lithium ion secondary battery according to claim  
2           2 wherein hole ratio of said porous adhesive resin layer is more  
3           than 35%.

1           4. The lithium ion secondary battery according to claim  
2 1 wherein the ionic conduction resistivity of said adhesive resin  
3 layer in which the electrolytic solution is kept is equal to or  
4 lower than the ionic conduction resistivity of said separator in  
5 which the electrolytic solution is kept.

1           5. The lithium ion secondary battery according to claim 1  
2 wherein a bonding strength between the positive electrode active  
3 material layer and the separator is equal to or larger than the  
4 bonding strength between said positive electrode active material  
5 layer and the positive electrode collector, and the bonding  
6 strength between the negative electrode active material layer and  
7 said separator is equal to or larger than the bonding strength  
8 between said negative electrode active material layer and the  
9 negative electrode collector.

1           6. The lithium ion secondary battery according to claim  
2 1 wherein the adhesive resin layer is made of a fluoride resin  
3 or a mixture which comprises a fluoride resin as the main  
4 component.

1           7. The lithium ion secondary battery according to claim  
2 6 wherein the fluoride resin is polyvinylidene fluoride.

1           8. A method of fabricating a lithium ion secondary battery,

2 comprising the steps of:

3        preparing a positive electrode by forming a positive

4        electrode active material layer on a positive electrode

5        collector;

6        preparing a negative electrode by forming a negative

7        electrode active material layer on a negative electrode

8        collector;

9        preparing an adhesive resin solution, by dispersing a

10      fluoride resin or a mixture containing a fluoride resin as the

11      main component in N-methylpyrrolidone;

12      coating said adhesive resin solution to at least one of

13      the surface of the positive electrode active material layer and

14      the facing surface of the separator and to at least one of the

15      surface of the negative electrode active material layer and the

16      facing surface of the separator;

17      fitting the positive electrode active material layer and

18      the negative electrode active material layer upon respective

19      surfaces of said separator;

20      evaporating said N-methylpyrrolidone from said adhesive

21      resin solution to form porous adhesive resin layers so as to bond

22      the positive electrode active material layer and the negative

23      electrode active material layer upon respective surfaces of said

24      separator to form a laminated body; and

25      supplying a lithium ion-containing electrolytic solution

26      to said laminated body.

1           9. The method of fabricating a lithium ion secondary  
2 battery according to claim 8 wherein said adhesive resin solution  
3 contains 3-10 parts by weight of a fluoride resin or a mixture  
4 containing a fluoride resin as the main component in N-  
5 methylpyrrolidone.

1           10. The method of fabricating a lithium ion secondary  
2 battery according to claim 8 wherein said step of evaporating  
3 comprises a step of heating.

1           11. The method of fabricating a lithium ion secondary  
2 battery according to claim 8 wherein said step of coating  
3 comprises a step of coating by using a bar coator.

1           12. The method of fabricating a lithium ion secondary  
2 battery according to claim 8 wherein said step of coating  
3 comprises a step of coating by using a spray gun.

1           13. The method of fabricating a lithium ion secondary  
2 battery according to claim 8 wherein said step of coating  
3 comprises a step of dipping the separator in an emulsified  
4 solution of the adhesive resin and then pulling it up.

**ABSTRACT**

To obtain a lithium ion secondary battery having excellent charge and discharge characteristics in which electric connection between electrodes can be maintained without requiring a strong armor metal case, so that it can be made into thin forms having  
10 large energy density. A positive electrode 3 prepared by bonding a positive electrode active material layer 7 to a positive electrode collector 6, a negative electrode 5 prepared by bonding a negative electrode active material layer 9 to a negative electrode collector 10 and a separator 4 which is arranged between  
15 these two electrodes and keeps a lithium ion-containing electrolytic solution are closely adhered by bonding the positive electrode active material layer 7 and the negative electrode active material layer 9 to the separator 4 by a porous adhesive resin layer 11, and an electrolytic solution is kept in through  
20 holes 12 formed in the adhesive resin layer 11, which communicate the positive electrode active material layer 7 and the negative electrode active material layer 9 with the separator 4.

【著者名】

Fig. 1

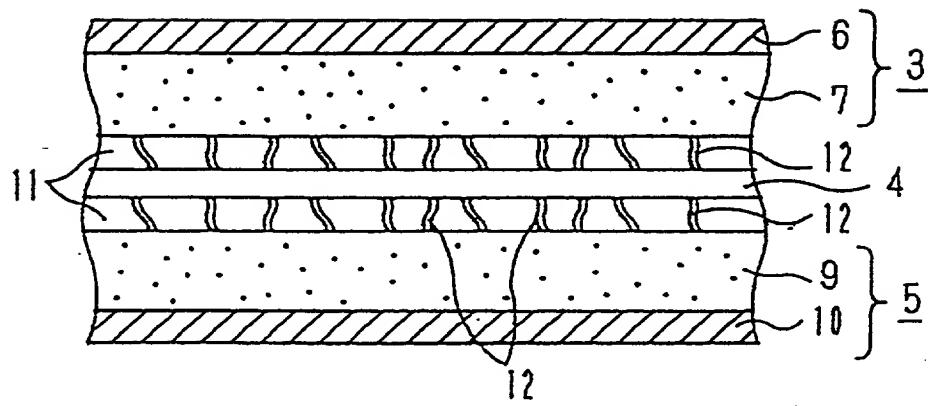


Fig . 2

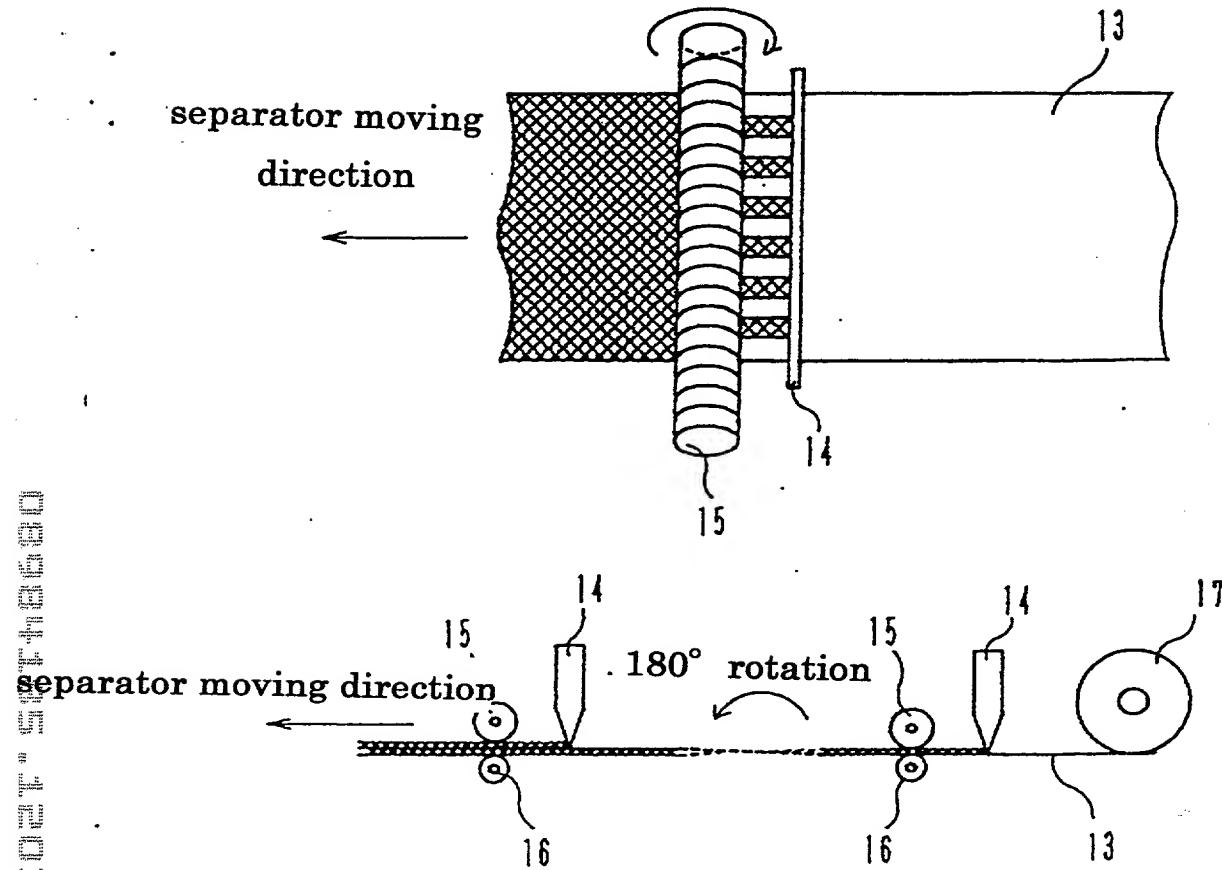


Fig 3

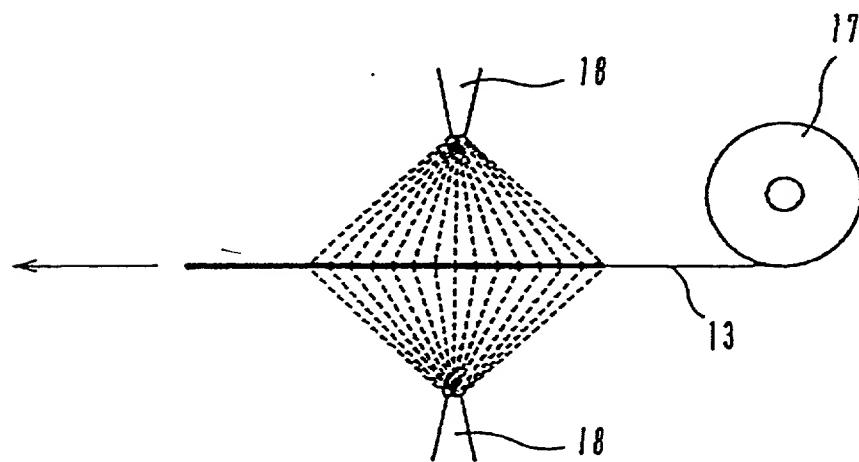
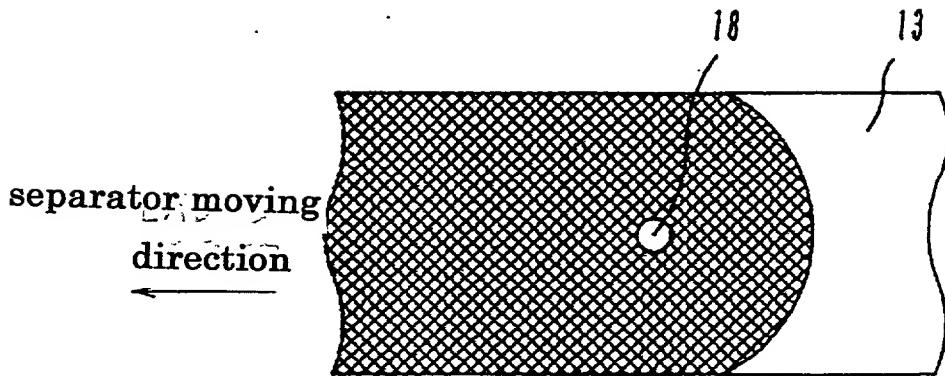


Fig. 4

separator moving  
direction

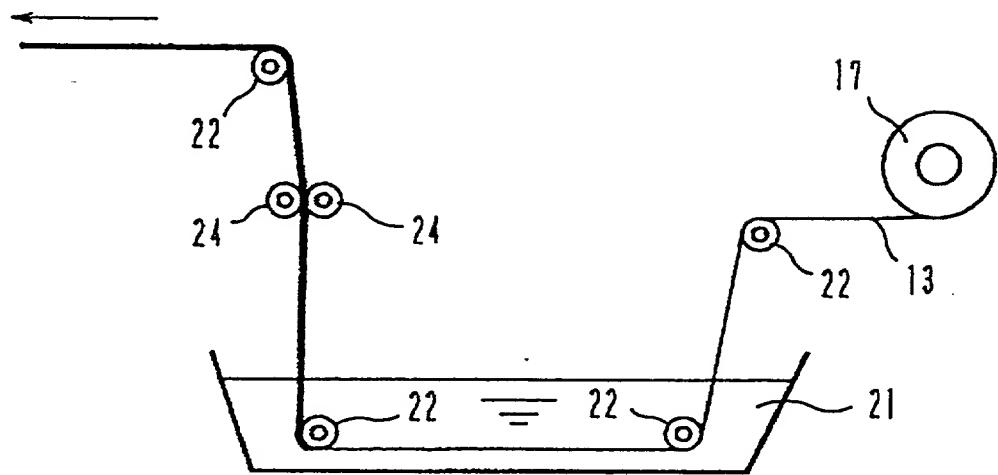


Fig. 5

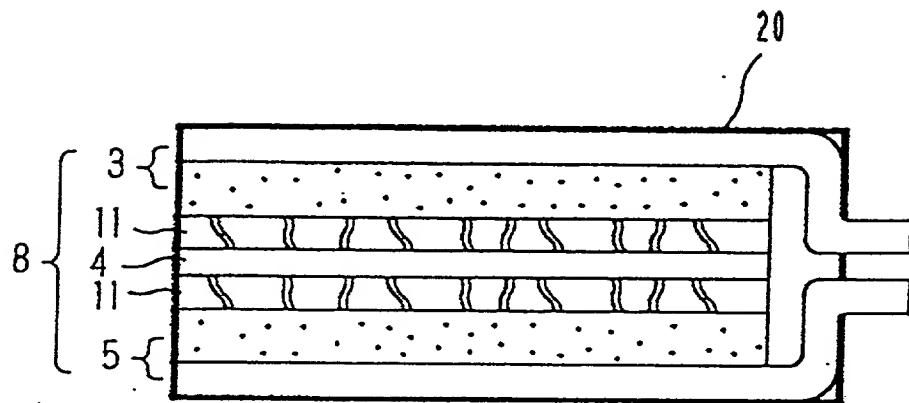


Fig. 6

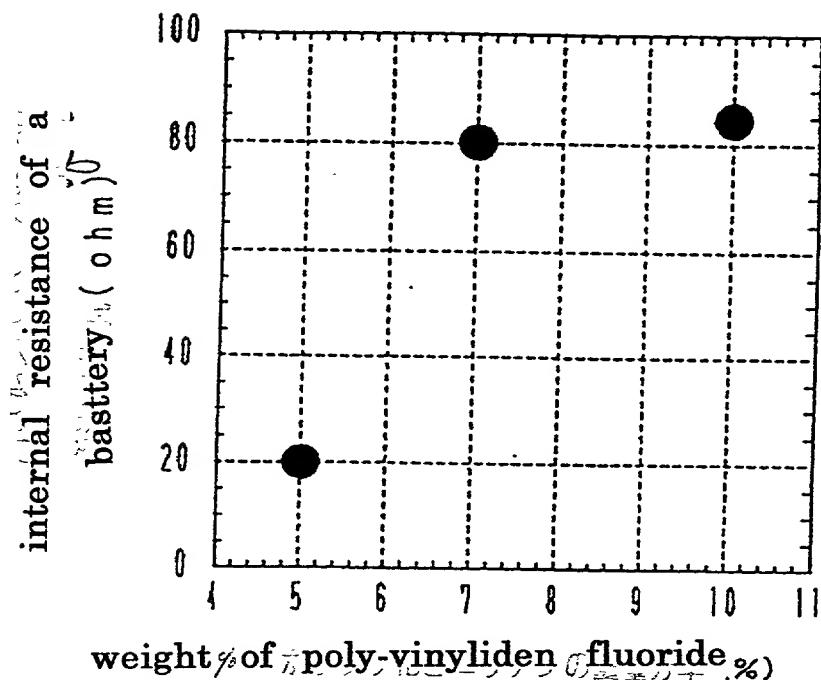
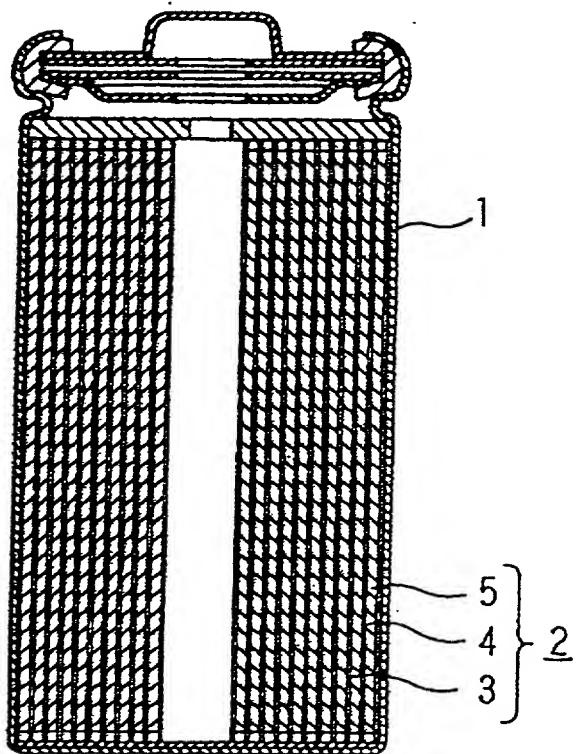


Fig. 7



# Declaration and Power of Attorney For Patent Application

## 特許出願宣言書 Japanese Language Declaration

私は、下欄に氏名を記載した発明者として、以下のとおり宣言する：

私の住所、郵便の宛先および国籍は、下欄に氏名に続いて記載したとおりであり、

名称の発明に関し、請求の範囲に記載した特許を求める主題の本来の、最初にして唯一の発明者である（一人の氏名のみが下欄に記載されている場合）か、もしくは本来の、最初にして共同の発明者である（複数の氏名が下欄に記載されている場合）と信じ、

\_\_\_\_\_  
\_\_\_\_\_  
\_\_\_\_\_

その明細書を  
(該当する方に印を付す)

ここに添付する。  
 \_\_\_\_\_ 日に出願番号  
第 \_\_\_\_\_ 号として提出し、  
\_\_\_\_\_ 日に補正した。  
(該当する場合)

私は、前記のとおり補正した請求の範囲を含む前記明細書の内容を検討し、理解したことを探述する。

私は、連邦規則法典第37部第1章第56条 項に従い、本願の審査に所要の情報を開示すべき義務を有することを認める。

As a below named inventor, I hereby declare that:

My residence, post office address and citizenship are as stated below next to my name,

I believe I am the original, first and sole inventor (if only one name is listed below) or an original, first and joint inventor (if plural names are listed below) of the subject matter which is claimed and for which a patent is sought on the invention entitled

### LITHIUM ION SECONDARY BATTERY AND METHOD OF FABRICATING THEREOF

the specification of which  
(check one)  
 is attached hereto.  
 was filed on \_\_\_\_\_ as  
Application Serial No. \_\_\_\_\_  
and was amended on \_\_\_\_\_ (if applicable)

I hereby state that I have reviewed and understand the contents of the above identified specification, including the claims, as amended by any amendment referred to above.

I acknowledge the duty to disclose information which is material to the examination of this application in accordance with Title 37, Code of Federal Regulations, §1.56.

## Japanese Language Declaration

私は、合衆国法典第35部第119条にもとづく下記の外国特許出願または発明者証出願の外国優先権利益を主張し、さらに優先権の主張に係わる基礎出願の出願日前の出願日を有する外国特許出願または発明者証出願を以下に明記する：

Prior foreign applications  
先の外国出願

<u>P.Hei.8-324032</u>	<u>Japan</u>	<u>04/December/1996</u>
(Number) (番 号)	(Country) (国 名)	(Day/Month/Year Filed) (出願の年月日)
_____	_____	_____
(Number) (番 号)	(Country) (国 名)	(Day/Month/Year Filed) (出願の年月日)
_____	_____	_____
(Number) (番 号)	(Country) (国 名)	(Day/Month/Year Filed) (出願の年月日)
_____	_____	_____

Priority claimed  
優先権の主張

<input checked="" type="checkbox"/> Yes あり	<input type="checkbox"/> No なし
<input type="checkbox"/> Yes あり	<input type="checkbox"/> No なし
<input type="checkbox"/> Yes あり	<input type="checkbox"/> No なし

私は、合衆国法典第35部第120条にもとづく下記の合衆国特許出願の利益を主張し、本願の請求の範囲各項に記載の主題が合衆国法典第35部第112条第1項に規定の態様で先の合衆国出願に開示されていない限度において、先の出願の出願日と本願の国内出願日またはPCT国際出願日の間に公表された連邦規則法典第37部第1章第56条項に記載の所要の情報を開示すべき義務を有することを認めます：

I hereby claim the benefit under Title 35, United States Code, §120 of any United States application(s) listed below and, insofar as the subject matter of each of the claims of this application is not disclosed in the prior United States application in the manner provided by the first paragraph of Title 35, United States Code, §112, I acknowledge the duty to disclose material information as defined in Title 37, Code of Federal Regulations, §1.56 which occurred between the filing date of the prior application and the national or PCT international filing date of this application:

(Application Serial No.) (出願番号)	(Filing Date) (出願日)
_____	_____
(Application Serial No.) (出願番号)	(Filing Date) (出願日)
_____	_____

(現 態) (特許済み、係属中、放棄済み)	(Status) (patented, pending, abandoned)
_____	_____
(現 態) (特許済み、係属中、放棄済み)	(Status) (patented, pending, abandoned)
_____	_____

私は、ここに自己の知識にもとづいて行った陳述がすべて真実であり、自己の有する情報および信ずるところに従って行った陳述が真実であると信じ、さらに故意に虚偽の陳述等を行った場合、合衆国法典第18部第1001条により、罰金もしくは禁錮に処せられるか、またはこれらの刑が併科され、またかかる故意による虚偽の陳述が本願ないし本願に対して付与される特許の有効性を損うことがあることを認識して、以上の陳述を行ったことを宣言する。

I hereby declare that all statements made herein of my own knowledge are true and that all statements made on information and belief are believed to be true; and further that these statements were made with the knowledge that willful false statements and the like so made are punishable by fine or imprisonment, or both, under Section 1001 of Title 18 of the United States Code and that such willful false statements may jeopardize the validity of the application or any patent issued thereon.

## Japanese Language Declaration

委任状：私は、下記発明者として、以下の代理人をここに選任し、本願の手続を遂行すること並びにこれに関する一切の行為を特許商標庁に対して行うことを委任する。  
(代理人氏名および登録番号を明記のこと)

POWER OF ATTORNEY: As a named inventor, I hereby appoint the following attorney(s) and/or agent(s) to prosecute this application and transact all business in the Patent and Trademark Office connected therewith. (list name and registration number)

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第2の共同発明者の氏名 (該当する場合)	Full name of second joint inventor, if any <b>Yasuhiro Yoshida</b>		
同第2発明者の署名	日付	Second Inventor's signature <i>Yasuhiro Yoshida</i>	Date November 28, 1997
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(第六またはそれ以降の共同発明者に対しても同様な情報および署名を提供すること。)

(Supply similar information and signature for third and subsequent joint inventors.)

## Japanese Language Declaration

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日付		Fifth Inventor's signature <i>Shigeru Aihara</i> Date November 28, 1997
住所		Residence <b>Tokyo, Japan</b>
国籍		Citizenship <b>Japan</b>
郵便の宛先		Post Office Address c/o MITSUBISHI DENKI <b>KABUSHIKI KAISHA, 2-3, Marunouchi</b> 2-chome, Chiyoda-ku, Tokyo 100 Japan
		Sixth Inventor's signature <b>Michio Murai</b> Date November 28, 1997
日付		Full name of sixth joint inventor, if any <i>Michio Murai</i>
住所		Residence <b>Tokyo, Japan</b>
国籍		Citizenship <b>Japan</b>
郵便の宛先		Post Office Address c/o MITSUBISHI DENKI <b>KABUSHIKI KAISHA, 2-3, Marunouchi</b> 2-chome, Chiyoda-ku, Tokyo 100 Japan

(第六またはそれ以降の共同発明者に対しても同様な情報および署名を提供すること。)

(Supply similar information and signature for third and subsequent joint inventors.)

**Japanese Language Declaration**  
(日本語宣言書)

第七共同発明者名(該当する場合)		Full name of seventh joint inventor, if any <i>Takayuki Inuzuka</i>	
第七発明者の署名	日付	Seventh inventor's signature <i>Takayuki Inuzuka</i>	Date November 28, 1997
住所	Residence Tokyo, Japan		
国籍	Citizenship Japan		
郵便の宛先	Post office address c/o MITSUBISHI DENKI KABUSHIKI KAISHA, 2-3, Marunouchi		
	2-chome, Chiyoda-ku, Tokyo 100 Japan		
第八共同発明者名(該当する場合)		Full name of eighth joint inventor, if any	
第八発明者の署名	日付	Eighth inventor's signature	Date
住所	Residence		
国籍	Citizenship		
郵便の宛先	Post office address		
第九共同発明者名(該当する場合)		Full name of ninth inventor, if any	
第九発明者の署名	日付	Ninth inventor's signature	Date
住所	Residence		
国籍	Citizenship		
郵便の宛先	Post office address		
第十共同発明者名(該当する場合)		Full name of tenth joint inventor, if any	
第十発明者の署名	日付	Tenth inventor's signature	Date
住所	Residence		
国籍	Citizenship		
郵便の宛先	Post office address		